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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=150 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.038$
$w R$ factor $=0.079$
Data-to-parameter ratio $=9.4$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## 2-C-Methyl-3,4-O-methylidene-D-arabinono-1,5-lactone

The relative stereochemistry at $\mathrm{C}-2$ of the title compound, $\mathrm{C}_{7} \mathrm{H}_{10} \mathrm{O}_{5}$, was determined by X-ray crystallographic analysis of the arabinonolactone, which adopts a boat conformation with a flagpole hydroxyl group. Its absolute configuration was determined by the use of D-erythronolactone as the starting material.

## Comment

Until recently, only linear carbohydrate chirons have been available as scaffolds for the synthesis of complex synthetic targets (Lichtenthaler \& Peters, 2004). However, the Kiliani cyanide reaction on ketohexoses (Hotchkiss et al., 2004; Soengas et al., 2005) affords versatile intermediates with carbon branches at C-2 of the sugar for the synthesis of imino sugars and complex sugar amino acids with non-linear carbon chains (Simone et al., 2005). The Kiliani reaction on hamamelose provides access to carbohydrates with a branch at C-3 (Parker, Watkin, Simone \& Fleet, 2006).

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(1)

(7)

(2)

(8)

(3)

(9)

(10)

Carbohydrate building blocks with a C-2 methyl group can be formed by the reaction of cyanide on 1-deoxyketoses, themselves prepared by the addition of organometallic reagents to sugar lactones (Hotchkiss et al., 2006). Thus, reaction of the isopropylidene-protected D-erythronolactone, (1), with methyl magnesium bromide followed by sodium cyanide gave the arabino-protected derivative, (2), as the only 1,5-lactone isolated (Punzo et al., 2005a). The potential of (2) as a route to sugar derivatives with a C-2 methyl group bearing a functional group at the tertiary centre is shown by its easy conversion to the branched arabinose, (3) (Punzo et al., 2005b), the quaternary ribo-azide, (4) (Punzo, Watkin, Jenkinson, Cruz \& Fleet, 2005), and the quaternary ribofluoride, (5) (Parker, Watkin, Mayes et al., 2006). The branched azidomethyl lactone, (6), has also been prepared from (2) and is a precursor to complex piperidine amino acids and iminosugars (Punzo et al., 2006).
In order to optimize the protecting group strategy for the synthesis of complex targets (and to investigate the diastereoselectivity of the Kiliani cyanide extension), the


Figure 1
The molecular structure of the title compound, with displacement ellipsoids drawn at the $50 \%$ probability level. H atoms are shown as spheres of arbitrary radii.


Figure 2
The crystal packing, viewed down the $a$ axis. Hydrogen bonds are shown as dashed lines.
formaldehyde acetal of D-erythronolactone, (7), was treated with methyl magnesium bromide to give the 1-deoxy-d-ribulose, (8). The Kiliani reaction of (8) with sodium cyanide gave a single diastereomeric product, (9), as the only 1,5-lactone isolated (Jenkinson et al., 2006). This paper shows, by X-ray
crystallography, that the arabinonolactone, (9), was formed in this reaction with none of the epimeric ribono diastereomer, (10), isolated.

The X-ray crystal structure determination shows that (9) is in a boat conformation (Fig. 1). The formation of (9) with the smaller hydroxyl group in the flagpole position may be due to the alternative product, (10), having the larger methyl group in the more hindered flagpole environment. The potential of (9) as a chiron is under investigation.

In the crystal structure, intermolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Table 1) link the molecules into zigzag chains extending along the $a$ axis (Fig. 2).

## Experimental

The title arabinono-1,5-lactone, (9), was obtained (Jenkinson et al.., 2006) by vapour diffusion of cyclohexane into a solution in ethyl acetate until crystals of a suitable size were formed (m.p. 373-375 K). $[\alpha]_{D}-126.0\left(c 1.0, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$, $\delta$, p.p.m.) : $1.67(3 \mathrm{H}, s, \mathrm{Me}), 3.04(1 \mathrm{H}, s, \mathrm{OH}), 4.25\left(1 \mathrm{H}, d, J_{3,4}=7.9 \mathrm{~Hz}, \mathrm{H} 3\right)$, $4.46-4.50(2 \mathrm{H}, m, \mathrm{H} 4, \mathrm{H} 5 \mathrm{a}), 4.82\left(1 \mathrm{H}, s, \mathrm{OCH}_{2} \mathrm{O}\right), 4.97\left(1 \mathrm{H}, d d, J_{4.5 \mathrm{~b}}\right.$ $\left.=1.9 \mathrm{~Hz}, J_{5 \mathrm{a}, 5 \mathrm{~b}}=12.0 \mathrm{~Hz}, \mathrm{H} 5 \mathrm{~b}\right), 5.17\left(1 \mathrm{H}, s, \mathrm{OCH}_{2} \mathrm{O}\right) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$, $\delta$, p.p.m.): 22.1 (Me), 68.7 (C5), 71.5 (C4), 72.2 (C3), $78.8(\mathrm{C} 2), 94.9\left(\mathrm{OCH}_{2} \mathrm{O}\right), 171.4(\mathrm{CO})$.

## Crystal data

$\mathrm{C}_{7} \mathrm{H}_{10} \mathrm{O}_{5}$
$M_{r}=174.15$
Orthorhombic, $P 2_{1} 2_{1} 2_{1}$
$a=6.8693$ (3) £
$b=7.0382$ (3) $\AA$
$c=15.7909$ (7) $\AA$
$V=763.45(6) \AA^{3}$

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.515 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \mu=0.13 \mathrm{~mm}^{-1} \\
& T=150 \mathrm{~K} \\
& \text { Needle, colourless } \\
& 0.50 \times 0.20 \times 0.20 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Nonius KappaCCD area-detector diffractometer
$\omega$ scans
Absorption correction: multi-scan (DENZOISCALEPACK;
Otwinowski \& Minor, 1997)
$T_{\text {min }}=0.882, T_{\text {max }}=0.974$
1726 measured reflections
1032 independent reflections
784 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.032$
$\theta_{\text {max }}=27.5^{\circ}$

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F^{2}\right)+(0.04 P)^{2}\right. \\
& \quad+0.03 P] \\
& \quad \text { where } P=\left(\max \left(F_{\mathrm{o}}^{2}, 0\right)+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.36 \mathrm{e}^{-3} \\
& \Delta \rho_{\min }=-0.34 \mathrm{e}^{-3}
\end{aligned}
$$

Table 1
Hydrogen-bond geometry ( $\AA,{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{O} 11-\mathrm{H} 1 \cdots \mathrm{O}^{\mathrm{i}}$ | 0.83 | 2.10 | $2.911(2)$ | 167 |
| Symmetry code: (i) $x+\frac{1}{2},-y+\frac{1}{2},-z+1$. |  |  |  |  |

A $[\sin (\theta) / \lambda]^{2}$ threshold of 0.01 was used to guard against the risk of including low angle reflections partially occluded by the beam stop. In the absence of significant anomalous scattering, 873 Friedel pairs

## organic papers

were merged and the absolute configuration was assigned from the known starting material. All H atoms were located in a difference map, but those attached to C atoms were repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their geometry $(\mathrm{C}-\mathrm{H}$ in the range $0.93-98 \AA$, and $\mathrm{O}-\mathrm{H}=0.825 \AA$ ) and isotropic displacement parameters $\left[U_{\text {iso }}(\mathrm{H})\right.$ in the range $1.2-1.5 U_{\text {eq }}$ of the parent atom], after which they were refined with riding constraints.

Data collection: COLLECT (Nonius, 2001); cell refinement: DENZO/SCALEPACK (Otwinowski \& Minor, 1997); data reduction: $D E N Z O / S C A L E P A C K$; program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: CRYSTALS (Betteridge et al., 2003); molecular graphics: CAMERON (Watkin et al., 1996); software used to prepare material for publication: CRYSTALS.

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